

Production and Decomposition Mechanisms of Reactive Oxygen Species by Red-tide Causing Phytoplankton — Case Study for Hydrogen Peroxide.

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H_2O_2 , *Chattonella antiqua*, *Heterocapsa circularisquama*, photochemical reaction, red-tide, H_2O_2 production and decomposition

In this doctor's thesis, the distribution of reactive oxygen species (mainly hydrogen peroxide (H_2O_2)) in the ocean was studied. I have studied the biological generation of H_2O_2 as well as the production by photochemical processes. Especially the phytoplankton that may cause harmful red tides and cultured fish and bivalves, in late spring to early summer in the Seto Inland Sea and other coastal seas in Japan were investigated for their ability of ROS production and decomposition by analyzing of natural red tide events.

Firstly, previous studies of production, distribution and decomposition of H_2O_2 in the environment mainly in the atmosphere and the ocean was summarized and on the basis of previous studies, the aim and significance of this study were described.

Secondly, the concentration and the behavior of H_2O_2 in the Hiroshima Bay seawater was investigated during 8 cruises in 1996 to 2002 (except 2000). H_2O_2 was characterized as higher concentrations at the surface water with decreasing trend with depth. The H_2O_2 concentration showed higher during the daytime (140-450 nmol L^{-1} at 5:00-19:00) than during the nighttime (85-260 nmol L^{-1} at 20:00-4:00) and suggested that H_2O_2 at the surface seawater was generated by photochemical reaction, so partly by biological production on the process of photosynthesis by phytoplankton. The correlation of H_2O_2 with environmental factors such as salinity was examined.

bloom period biological production may be the dominant for H_2O_2 generation. In addition, it is clear that some organisms associated with red tide in Japan and other countries have the specific mechanism of H_2O_2 production and decomposition. Considering significant fishery damage reported by this species, further clarification of production and decomposition processes of ROS is needed.